

Chemistry Department

Chem. 233 (Exam II)

Reg. number: Section: Seat no.:...

Q1 (22.5 points): Circle the correct answer in each of the following:

- The correct name of CH_3 -CH-CH=CH- CH_3 is?
 - a- 2-methyl-3-pentene
- b- 2-methoxy-3-pentene
- c- 1-methoxy-1-methyl-2-butene
- d- 4-methyl-2-pentene

e 4-methoxy-2-pentene

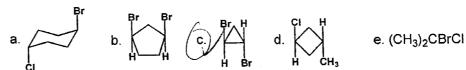
Ξ The most reactive toward substitution with CH₃OH is

Z The most reactive toward substitution with CH3O Na is

 $\not\equiv$ The solvent with the highest rate of reaction in S_N1 type reaction is?

- b- ether
- c-benzene
- d- hexane
- e- all are the same

Ξ Which of the following compounds is chiral?





 Ξ Which of the following groups have the highest priority according to priority rules?

$$(H = 1; C = 12; N = 14; O = 16)$$

≡ Which of the following is the <u>enantiomer</u> of R-2-butane?

$$(H_3 \xrightarrow{CH_2(H_5)} H_3C \xrightarrow{CH_2CH_3} H_3C \xrightarrow{H} P_3C \xrightarrow{H}$$

- a. only I
- b. only II (c. only III
 - d. I and II
- e. I and III

≡ Which of the following alkenes has Z-configuration?

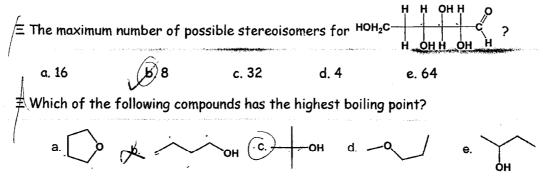
= Which of the following compounds is dextrorotatory?

≡ Which of the following is not true about enantiomers?

- a. They have the same melting and boiling points.
- b. They have the same absolute value of specific rotation.

2 1:1 mixture is optically active.

- d. They could be separated using a chiral reagent.
- e. none of the above.



≡ Which of the following is the strongest acid?

La- enantiomers b- diastereomers c- meso compound d-identical e-not related

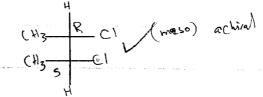
 \equiv The IUPAC name of this compound HO is

- a- (R)-3,5-dihydroxypentane
- b-(5)-3,5-dihydroxypentane
- c- (R)-1-hydroxy-3-pentanol

d-(S)-1,3-pentandiol (e-(R)-1,3-pentandiol

Q2 (10 Pt) Draw the structure of each of the following

- The Fischer projection of the <u>optically inactive</u> 2,3-dichlorobutane.



A stereoisomer of
$$CH_3$$
 $CH=CHBr$ which has an opposite sign of rotation.

$$CH=CHBr$$
 which has an opposite sign of rotation.

- C₄H₇Cl which exhibits both cis/trans isomerism and optical activity.

- The E-isomer of Br₂C=CH-CH=CHBr

- $C_4H_{10}O$ alcohol which is <u>not</u> oxidized by CrO_3 .

Q3 (15 pt) Show the synthesis of each of the following from the designated starting material, use any other needed reagents.

$$5$$
 $-CH_2-SH$

From

 $-CH_2-OH$
 $-CH_2$

CH₃—CH—CH₃ From CH₂—CH—CH

5 D

1- CH₂=CH -CH₃ HBCD CH₃ CH -Ch₃

Br

2- CH₃ CH CH₃ M9 D CH₃ CH -Ch₃

Br

R₃ CH -CH₃

Br

R₃ CH -CH₃

R₄ CH -CH₃

R₅ CH -CH₅

R₅ CH -CH

HO—CH2-CH2—OCH3 From HO—CH2-CH3 PBF2 (N3 CH2B)

(CH3 (N2 B) + (N3 CH2 OH) A — CN3 (N2 OH) PBF2 (N3 CH2B)

(CH3 (N2 B) + (N3 CH2 OH) A — CN3 (N2 OH) PBF2 (N3 CH2B)

(CH2 CH2 CH2 III)

(CH2 CH2 CH2 III)

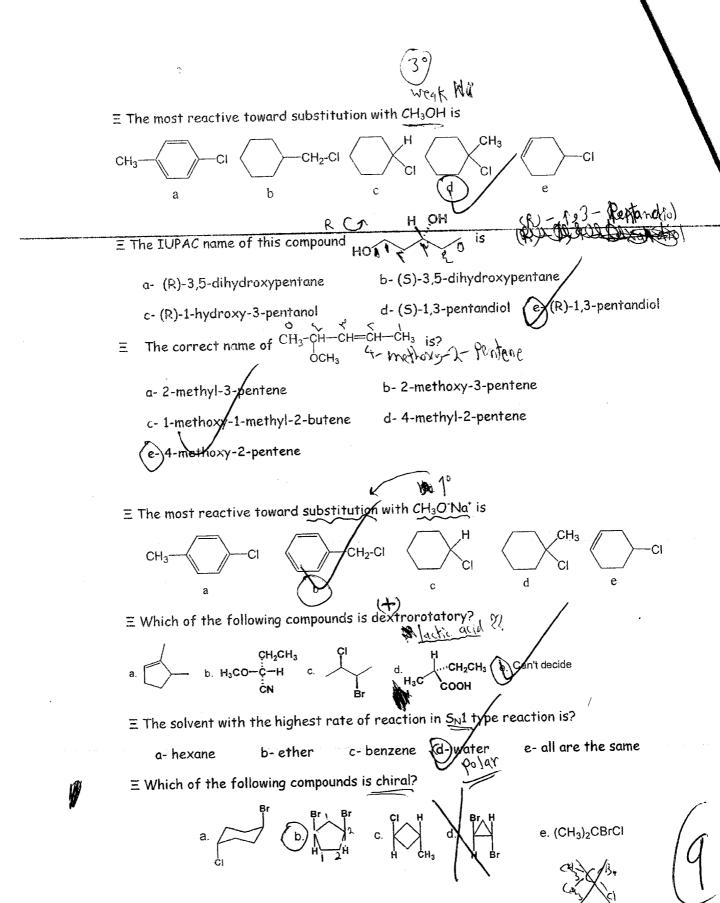
(CH2 CH2 CH2 III)

(CH2 CH2 CH2 OCH3 III)

4

Q4 (28 points): Complete the following equations by writing $\frac{\text{product(s)}}{\text{show the}}$ stereochemistry where appropriate: Write the name of the mechanism by which the product is formed $(S_N 1, S_N 2, E1, E2)$ for the first four reactions.

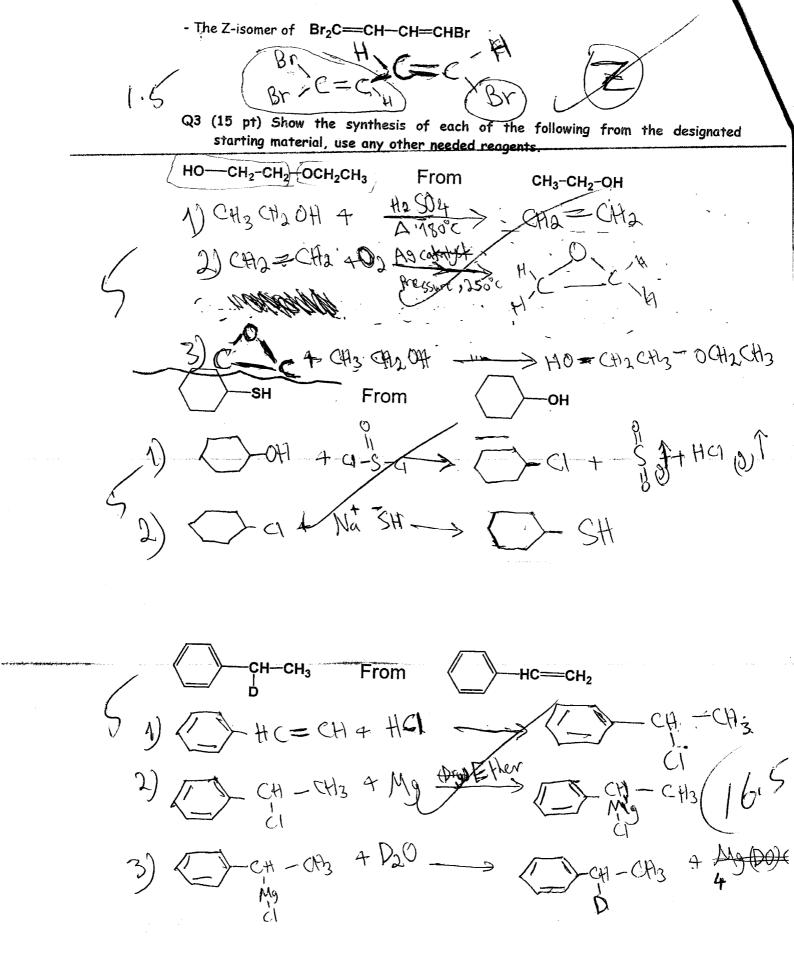
Chemistry Department Chem 233 (Exam II) 29/12/2013	
Name: Seat no:	
Q1 (22.5 points): Circle the correct answer in each of the following:	
Ξ Which of the following is <u>not true</u> about enantiomers?	مزسد
a. They have the same melting and boiling points.	/ -
b) 1:1 mixture is optically active.	
c They have the same absolute value of specific rotation.	
d. They could be separated using a chiral reagent.	
e. none of the above.	
Ξ Which of the following is the strongest acid?	
a. OH OH OH OH OH OH OH OH	
E The maximum number of possible stereoisomers for HOH2C HOHHOHH?	ζ
a. 64 b. 4 c. 32 (d.)8 e. 16	
≡ Which of the following compounds has the highest boiling point?	
а. О О О О О О О О О О О О О О О О О О О	
E This pair of compounds H Br Br H are	
a- identical b- diastereomers c- meso compound	
d enantioners e-not related	
C: P:	
2 week to truly	
the second secon	



 Ξ Which of the following groups have the highest priority according to priority rules? b. -CH=C-CH₃ c. -CH=CH-CH₃ d. -C=C-CH₃ e. -C≡N

C,C,H CH₃ C, €, H Which of the following is the enantiomer of R-2-butane? c. only III ≡ Which of the following alkenes has E-configuration? Q2 (10 Pt) Draw the structure of each of the following $^{\circ}$ - $C_4H_{10}O$ alcohol which is not oxidized by CrO_3 . - A stereoisomer of CH3 CH=CHBr which has an opposite sign of rotation. CH₃ \rightarrow CH₃ \rightarrow CH₃ \rightarrow CH₃ \rightarrow CABY

- C₄H₇F which exhibits <u>both</u> cis/trans isomerism and optical activity. - The Fischer projection of the optically inactive 2,3-dibromobutane.



Q4 (28 points): Complete the following equations by writing $\frac{\text{product(s)}}{\text{show the}}$ stereochemistry where appropriate: Write the name of the mechanism by which the product is formed (S_N1 , S_N2 , E1, E2) for the first four reactions.

4

