

Nedical Committee The University of Jordan

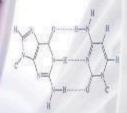
## SLIDE O SHEET

SLIDE: 10



DR.NAME: Nayef Kradsheh

## Biochemistry



Majida Al-Foqaraa'

7 Production of reactive Drygen species during the phagocytic Respiratory burst by activated macro entrophils of easin ophils HOCL - hypochlorous ltoxin-helogen Respiratory , O<sub>2</sub> Activation NADPH INDS and OX ... burst itiates the 0 NADPH oxidase - 4 SH - centers eg. Fe burst ox. de garboxylation 9 NADP\* 02 Deficiency м0 ox. deamination spontaneous reaction or 2 ulo matesis SOD breaking pep Bacterium H20 nitrite Fe<sup>2+</sup> Fenton reaction ONOD Per Igenerate RNOS Fe<sup>3+</sup> HOCL 3 -CI HOC myeloperoxidase chloru > OH-Bacterium ONOOP . Invagination of neutrophil's cytoplasmic membrane OH + HOCL attacks bacterial cells Peroxymi HON lysis } + Respiratory 30 202 + Consumption 102 NO3 safe udical fration ajent) (ni

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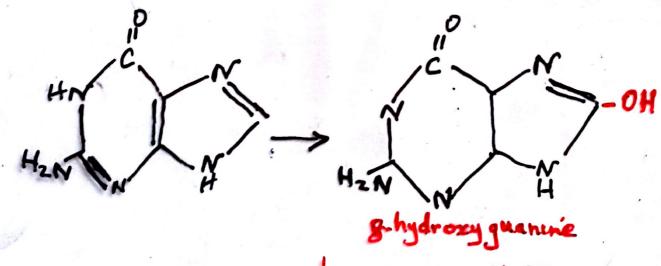
Synthesis of Nitric Oxide (NO) 9 NH2 NH2 NADP c = 0C=NH2 HOPH (CH2)3 NOSYNFLOSE HCMH3 COO citrulline HENHS COO Arginine Relaxes smath muscle . GTP Cyclose CGMP NO(+) CGMP activates PKG PKG phosphorylates Ca protein channels which decrease Catentry to muscle cells which decrease Ca-calmo dulin of Eherefore decreasing muscle contraction and favoring relaxation of vascular smusc Prevents platelet aggreg ation Neurotransmitter in Brain Mediates tumoricidal and bactericidal actions & macrophages

Net oxidative and free radical damage of RNOS :-

. Inhibition & large number of euzymes :- attacking met 4-stgr 901

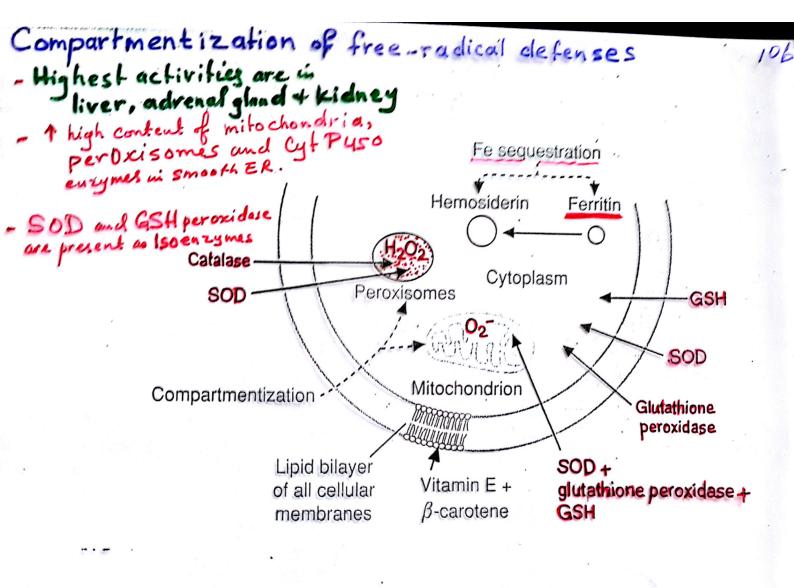
- . mitschondrial lipid peroxidation
- . In hibition of electron transport chain and energy depletion
  - Single- and double-stranded breaks in DNA

Modification of bases in DNA



Cellular Defence Agamist 0 toxicity:-Primary Antioxidants - Antioxidant Enzymes SOD; Catalase, GSH peroxidase, Highest conc. in liver, adrenal + kidney (high contents perozisomal + mit) a GSHreductose. > On the H2O2 to OH to H2O 01 hydrogen peroxide eductase NADPH 655 G a antiorida 2H20 2H20+02 Gly Cys GSH Glu , G55G GSH + GSH -

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- Secondary Antioxidants 12-A. Dietary :-() Vitamins Vitamin E (tocopherol) Vitamin C B- carofenes (2) other dietary antioncidents B. Endogenous antioncidants Repair Mechanism of Repair Mechanism of JAR, onidized fatty acids & DNA, onidized fatty acids & membrane lipids and muldized amining acids C. D. Comportmentation e's peroxisomes, familie for fet Scanned by CamScanner

Other Dietary Antioniclants Flavonoids (Polyphenolic comprunds) Green-ten Choclate red wine Fruits skin ward Possible (inhibition & Di production e.g. X.O.) Functions (xenthine excidence) Chelate Fe + Cu maintenance Vit E Endogenous AnHoxidants Uric acid GSH melatonin Bilirubin lipoic acid Ubiquinone (Co Qio) ·Catechins :- strawberries, greent black tea Some flavenoids: ·Kaempferol:-brussel sprouts & apple guercetin:-beans, onions, apples and fruit. skin Espicatechin: - Cocoa, red wine

Vitamin Antioncoants.

Vit. E: most widely Chain - breaking distributed antioxidant antioxidants -Terminates free Gitte radical lipid peromidation **d**-Tocopherol -L00. . Vit. E donate single e OOH . Constensido occupt e from lipid perozy radiculs . Vit C accepts single H<sub>3</sub>C e from Di, H2 02, BH', HOLL & perory. Tocopheryl radical radicals 100. . Vit C regensate the reduced form & Vil. E O Phylyl

- sale physiological

role is to quench free radical reactions

## Tocopheryl quinone

Fig. 21.15. Vitamin E. Vitamin E terminates free radical lipid peroxidation by donating single electrons to form the stable, fully oxidized tocopheryl quinone. Of the eight or more different tocopherols that comprise vitamin E, a-tocopherol, shown here, is the most common in the diet.

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12% Vit. HO H HO +e<sup>-</sup> он 0 ÔН ÕН Ascorbate Ascorby radical Dehydro-L-ascerbic acid B-corotenes - precursor & vit. A Corokenoids :-L00. OOL All-trans-ß-carotene A lipid-peroxy product Zea xan thine (macular campenoids) Frich in dark green leafy regetables Lutein 6 -----